UDC 547.944/945

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We have investigated the plant Magnolia soulangeana Soul-Bod (a hybrid of the Chinese species M. vulan and M. obovata), family Magnoliaceae [1], collected in September, 1974, in the botanical gardens of the Academy of Sciences of the Uzbek SSR, Tashkent.

Ordinary chloroform extraction of the leaves yielded 0.105% and of the stems 0.14% of total alkaloids. From the mixture of bases obtained from the stems of this plant by chromatography on a column of silica gel we isolated liriodenine [2] and an optically inactive base (I) with the composition $C_{18}H_{11}NO_4$ with mp 265-267°C (from methanol), R_f 0.56 in the

benzene-ethanol (4:1) system in a thin layer of silica gel. The base dissolves well in acid, sparingly in chloroform, ethanol, and methanol, and is insoluble in alkali. The UV spectrum of (I) $[\lambda_{\max}^{\text{EtOH}}, 249, 270, 309, 349 \text{ nm} (\log \epsilon 4.21; 4.08; 3.72; 3.84), \lambda_{\max}^{\text{EtOH+HC1}} 259, 280,$ 379 nm (log ε 4.07; 3.99; 3.73)] is typical for alkaloids of the 12-oxodibenzo [de, g] quinoline series [3]. The IR spectrum of the base has maxima at (cm⁻¹): 2850, 1265 (-OCH₃), 1605 (aromatic nucleus), 1060, 970 ($-OCH_2O-$), and 1650 (>C=O); there are no absorption bands of -OH and >N-H groups.

The mass spectrum of (I) shows strong peaks of ions with m/e M⁺ 305 (100 %), 290 (M-15)⁺, 275 (M-30)⁺, 262, 234, 206, 204, 176, 175, 149 and M⁺⁺ 152.5, which are characteristic for alkaloids of the liriodenine type [4].

In the NMR spectrum of the alkaloid (JNM-4H-100/100 MHz in CF₃COOH, internal standard HMDS, δ scale) two singlets from protons of methoxy (3.68 ppm, 3H) and methylenedioxy (6.22 ppm, 2H) groups appear distinctly, and in the weak field there are the signals of six aromatic protons. A one-proton singlet at 7.16 ppm, analogous to the singlet observed in the spectrum of liriodenine, corresponds to the C, aromatic proton [2], and two one-proton doublets at 8.17 ppm and 8.35 ppm (J = 7.0 Hz) relate to the two protons at C₈ and C₉. Analysis of the signals of the other three protons: doublet at 8.05 ppm (J = 8.5 Hz), ortho

quartet with its center at 7.02 ppm (J = 8.5 Hz, J = 2.5 Hz), and a signal at 8.45 ortho ppm show that they belong to a 1,2,4-substituted benzene ring. Hence, the methoxy group in (I) is located in ring D at C_2 or C_3 .

The presence of the methylenedioxy group in the C_5-C_6 position in the aporphine ring [5] and the nonidentity of the alkaloid (I) with lanuginoside [6, 7], which we have also isolated from the plant Liriodendron tulipifera (family Magnoliaceae) indicate that (I) is oxolaureline - 3-methoxy-5,6-methylenedioxy-12-oxodibenzo [de, g] quinoline.

Substance (I) has been synthesized previously [8], but we are the first to have found it in a plant.

LITERATURE CITED

1.

Flora of Uzbekistan [in Russian], Vol. II, Tashkent (1953), p. 516. A. Abdusamatov, R. Ziyaev, and S. Yu. Yunusov, Khim. Prirodn. Soedin., 112 (1974). 2.

3. M. A. Buchanan and E. E. Dickey, J. Org. Chem., 25, 1389 (1960).

Institute of the Chemistry of Plant Substances, Academy of Sciences of the Uzbek SSR. Tashkent Agricultural Insitute. Translated from Khimiya Prirodnykh Soedinenii, No. 4, pp. 528-529, July-August, 1975. Original article submitted February 11, 1975.

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- J. R. G. Bick, J. H. Bowie, and G. K. Douglas, Aust. J. Chem., 20, 1403 (1967). 4.
- 5. S. Yu. Yunusov, Dokl. Akad. Nauk UzSSR, No. 8, 12 (1948).
- 6.
- 7.
- S. M. Kupchan, M. J. Suffness, and E. M. Gordon, J. Org. Chem., <u>35</u>, 1682 (1970). S. K. Talapatra, A. Patra, B. Talapatra, Chem. Ind. (London), 1056 (1969). T. Govindachari, N. Viswanathan, S. Narayanaswani, and B. R. Pai, Indian J. Chem., <u>8</u>, 8. 475 (1970).